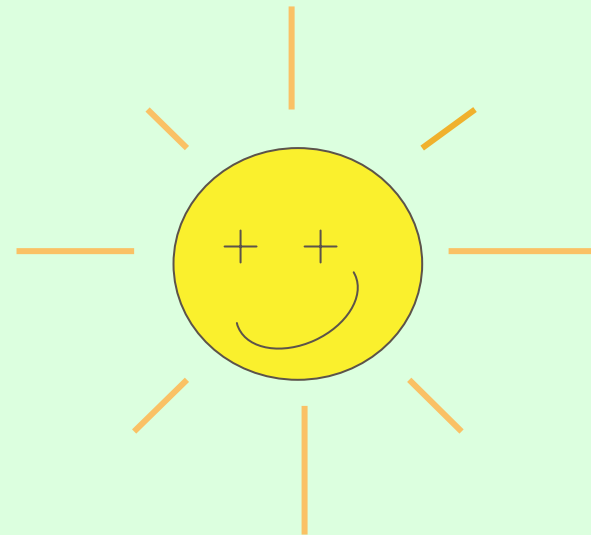
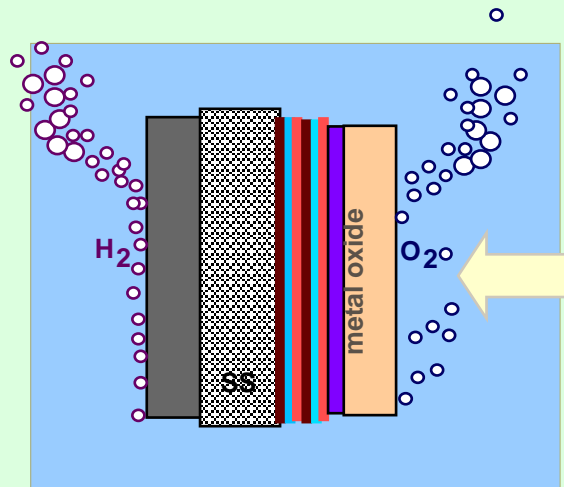




To build a photoelectrochemical (PEC) system that produces hydrogen fuel directly from water using sunlight as the energy source.

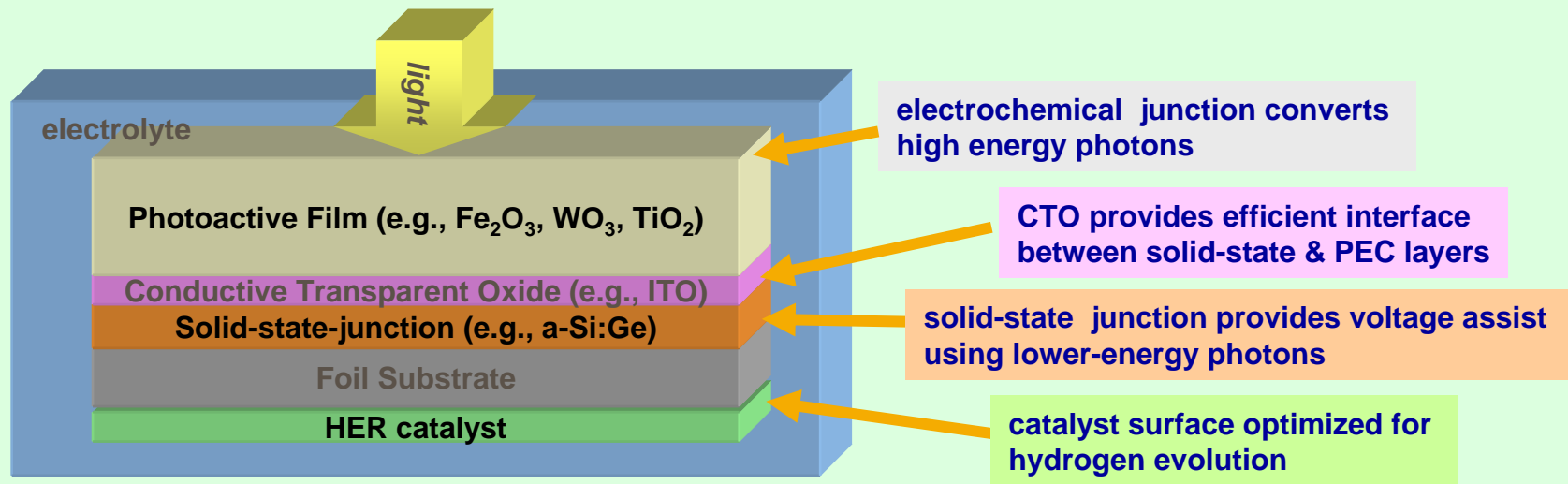


Approach: development of a multi-junction monolithic photoelectrode using low cost thin films materials

The Hybrid Photoelectrode



2



UH-developed integrated multijunction electrochemical device to provide sufficient photo-generated current & voltage for efficient water splitting

Oxide layer is photoactive and corrosion resistance



tungsten and titanium oxides are photoactive, but their high bandgap limits photocurrent

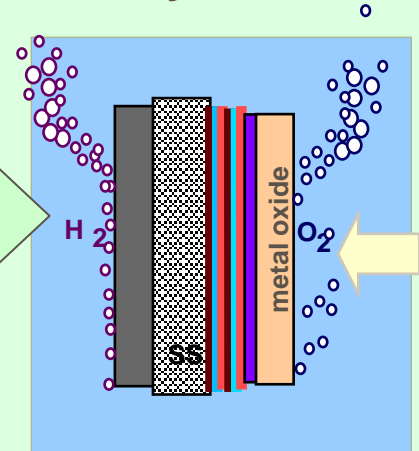
WO₃ or TiO₂ with reduced bandgap
(doping, surface mod, etc)

iron oxide has near-optimal bandgap, but recombination losses limit photocurrents

Fe₂O₃ with improved electronic properties
(doping, surface mod, etc)

Integration with optimized solid state materials/devices

Viable PEC System



Other materials may arise through combinatorial discovery, etc....

Why WO₃ ?



- Conduction band minimum well aligned with H⁺/H₂ level
- Bandgap wide enough to produce photoelectrolysis of water
- Good chemical stability demonstrated in 1 Normal H₃PO₄
- Low cost material, environmentally friendly

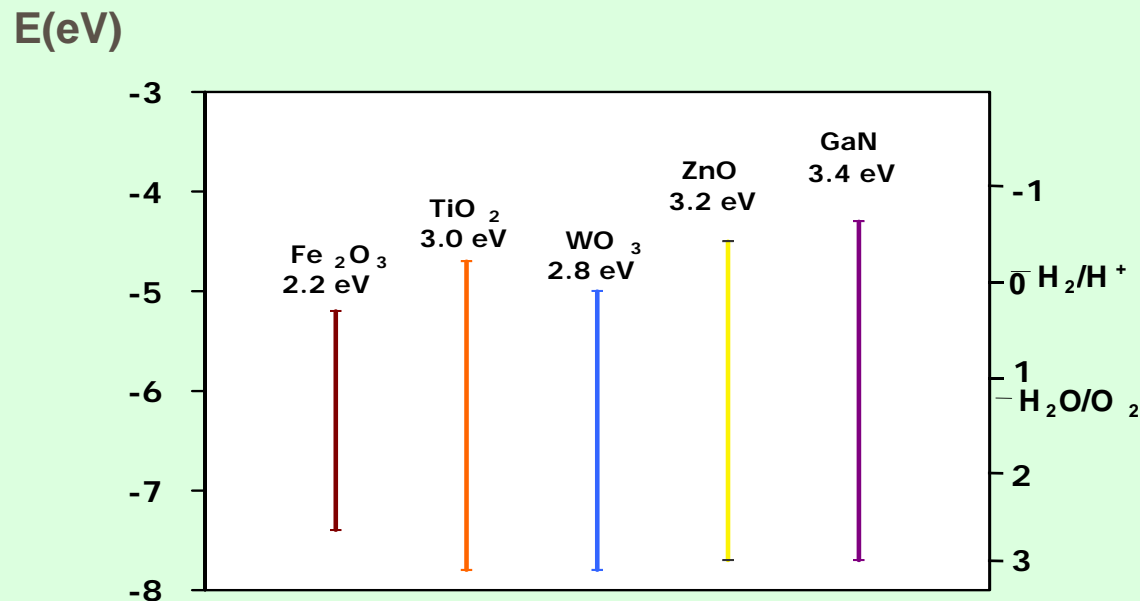
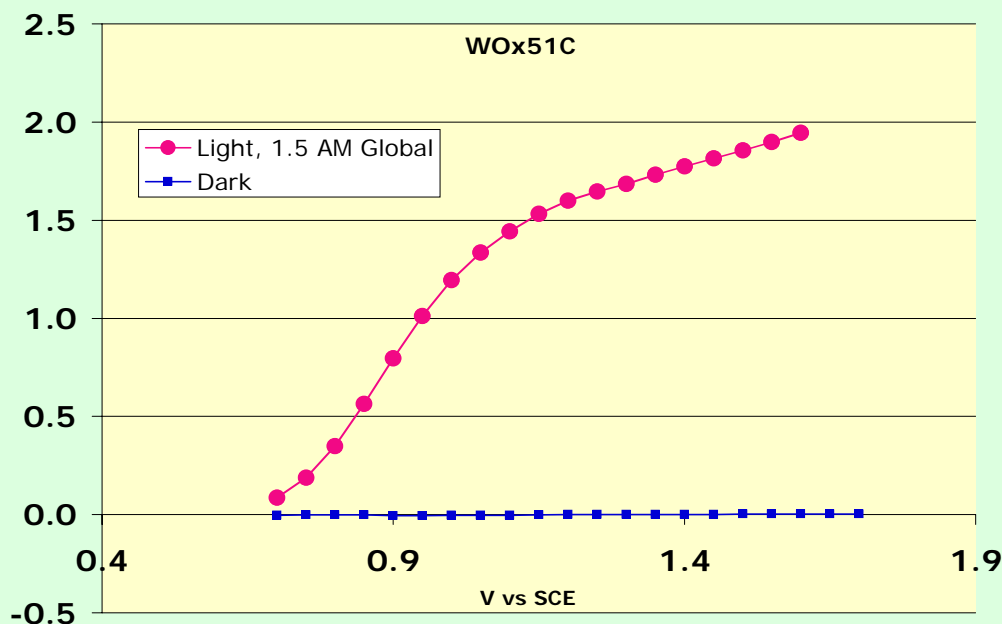
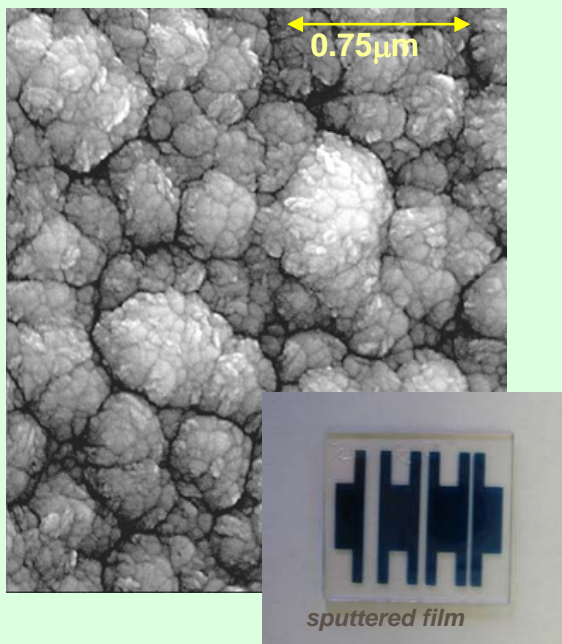


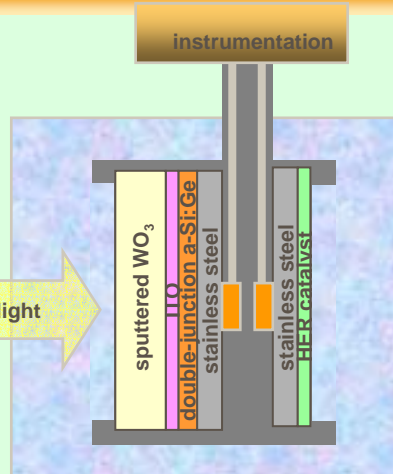
Table from T. Lindgren, 2001 (pH=1)



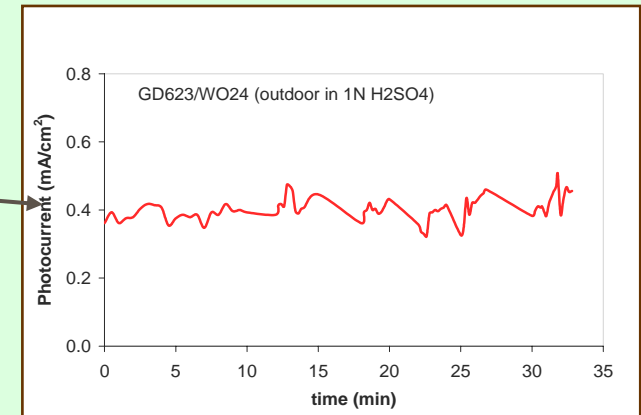
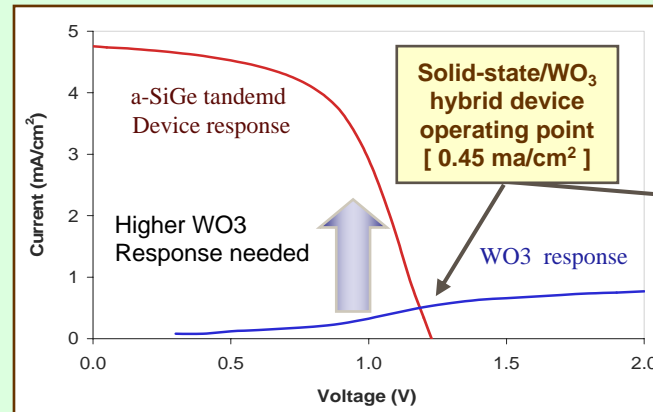
- WO₃ thin films are produced via reactive sputtering from a tungsten target
- To date, films deposited at low temperature (200°C) with photocurrents up to 2 mA/cm² under 1 sun illumination in phosphoric acid



Early Prototype Photoelectrodes



outdoor hydrogen production current



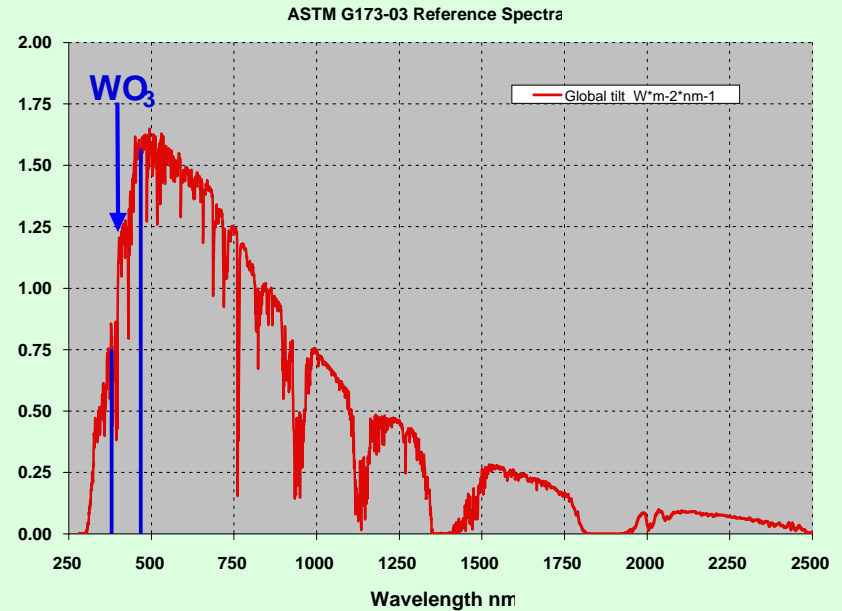
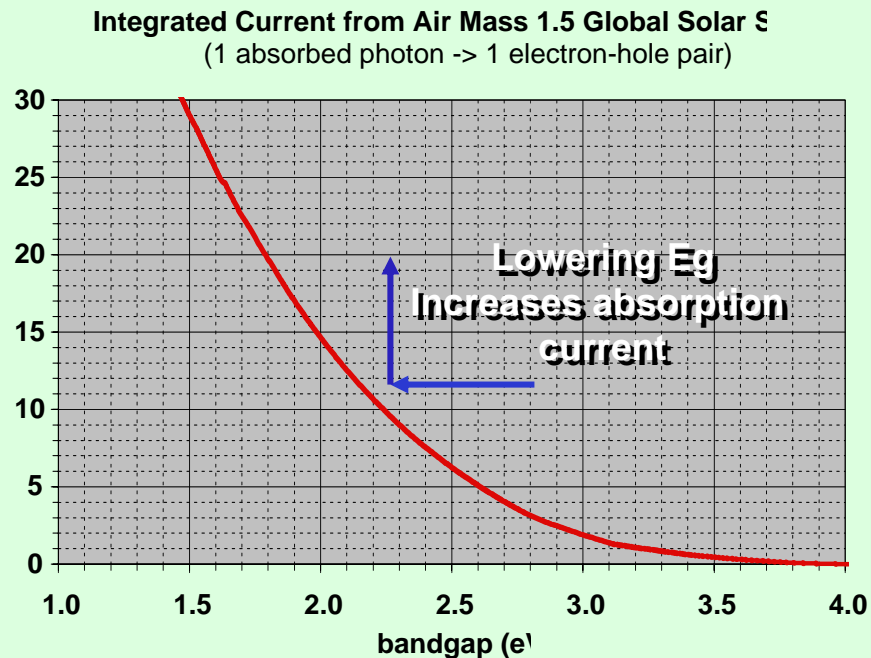
- Currents up to 0.5 mA/cm² in 1-sun outdoor tests were achieved in **unbiased conditions** (0.7% Solar to Hydrogen Efficiency)
- Stable operation in 1N H₂SO₄ was measured up to ten hours

Improved metal-oxide properties are necessary to enhance STH efficiency conversion

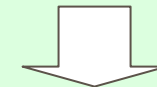
WO₃ Bandgap versus Current Absorption



Because of its high bandgap ($E_g \sim 2.6\text{-}3.2$ eV) WO₃ mainly absorbs in the near ultraviolet and blue region of the spectrum ($\sim 470\text{-}380$ nm).



To include a wider region of the solar spectrum, and therefore increase PEC performance, the WO₃ bandgap must be lowered.



Doping as one of the pathways for bandgap reduction

Why Nitrogen Doping

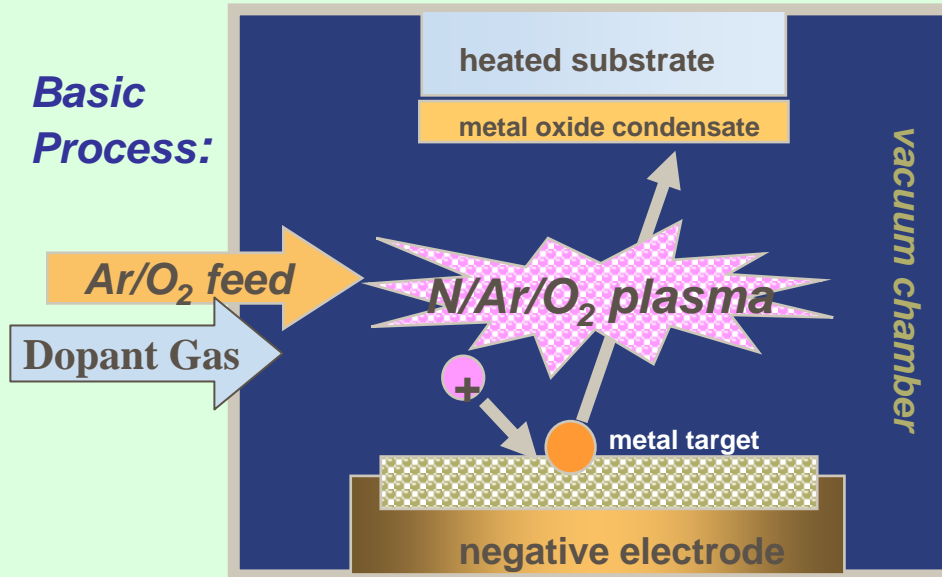


The VB of WO_3 is mainly produced by the O 2p electrons and the CB is mainly generated by the W 5d electrons (Detraux et al. 1997, etc...).

- Anion impurities substituting on O sites will introduce purity defects bands above the VB of WO_3
- Cation impurities substituting W may introduce impurity bands below the CB.

Impurities that can be considered for bandgap reduction include N, S, C, P, As, Cu...

Asahi et al., 2001 (on N-doped TiO_2): “Substitutional doping of N was the most effective because its p states contribute to the bandgap narrowing by mixing with the O 2p states”



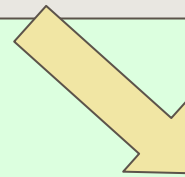
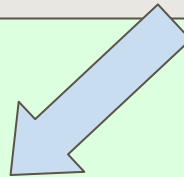
- Un-doped WO_3 films are produced via 'low-temperature' reactive r.f. sputter from a metallic target, using Argon and Oxygen gas.
- Nitrogen containing gas (N_2 or NH_3) are introduced along with Argon and Oxygen, to produce nitrogen doped tungsten trioxide films.

Important process parameters influencing film properties are: ambient pressure, oxygen partial pressure, substrate temperature, r.f. power, geometry, substrate material.



Standard baseline

Ambient Pressure: 10 mTorr.
Oxygen partial pressure: 12%
Substrate Temperature: 200C
r.f. Power: 300W



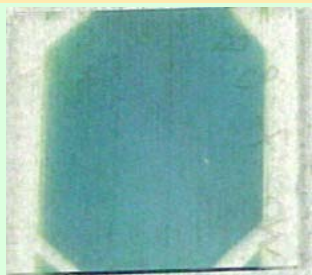
N₂ Doping

Nitrogen partial pressure → 3.5%
Nitrogen partial pressure → 1.8%

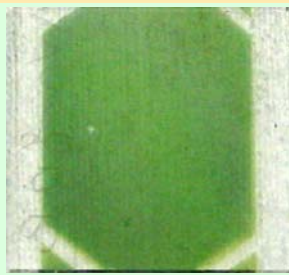
NH₃ Doping

2.3% ← Ammonia partial pressure
1.8% ← Ammonia partial pressure

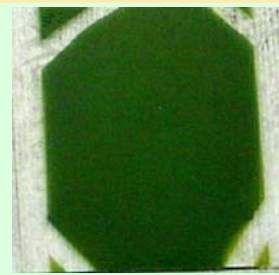
11 N₂ Doping: Optical & Electronic Properties



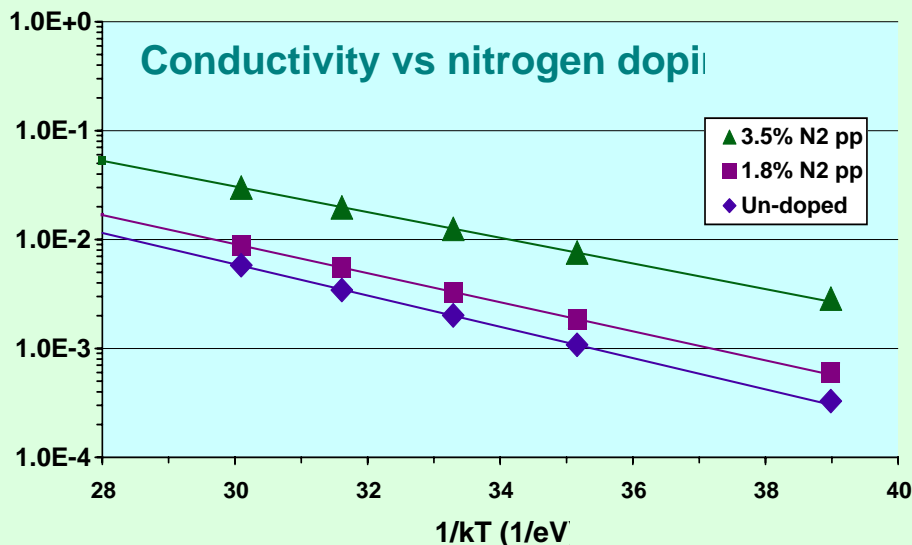
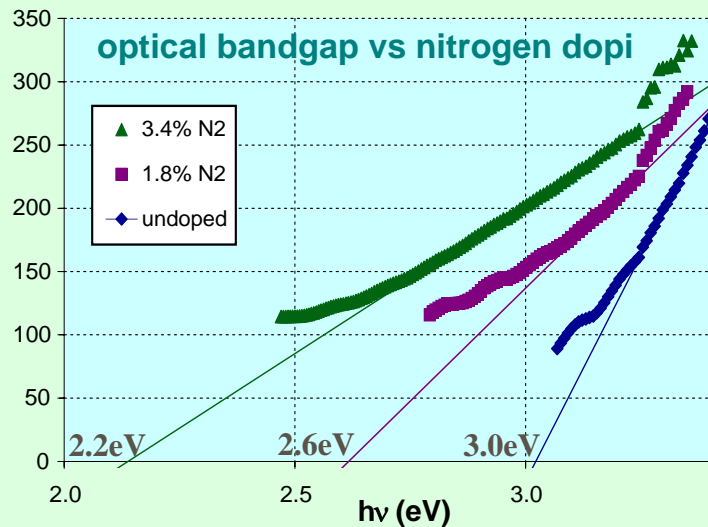
N₂ : 0%



N₂ : 1.8%

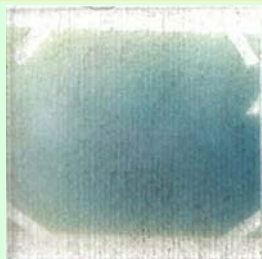


N₂ : 3.5%



As N₂ doping \uparrow \Rightarrow the bandgap \downarrow , the conductivity (σ) \uparrow , and the activation energy (slope) \downarrow

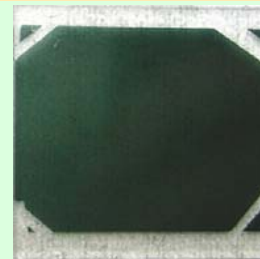
12 NH₃ Doping: Optical & Electronic Properties



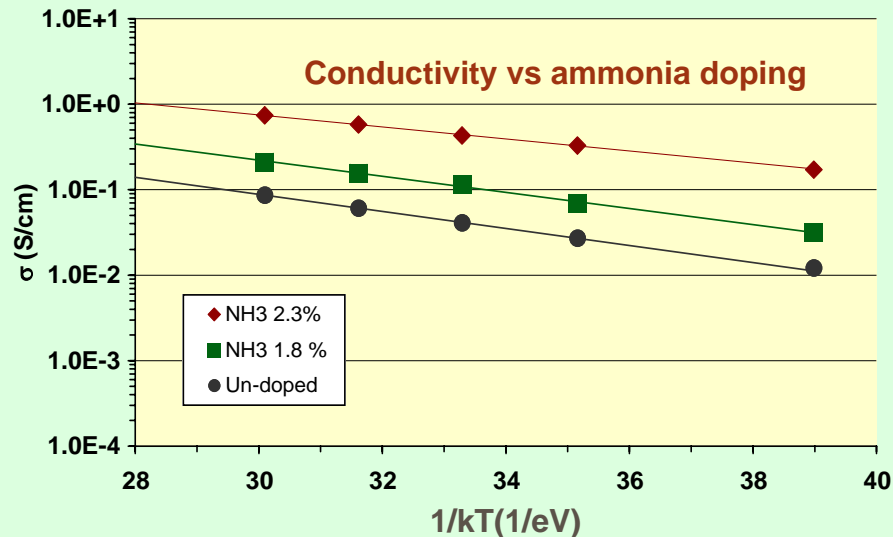
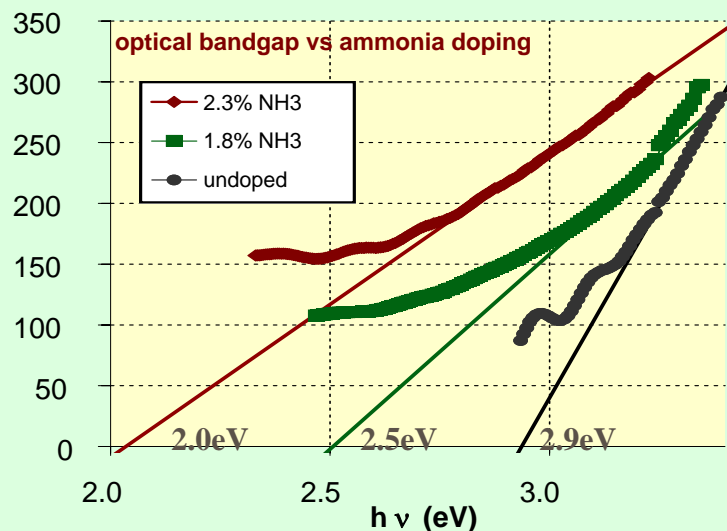
NH₃ : 0%



NH₃: 1.8%



NH₃ : 2.3%



As NH₃ doping ↑ ⇒ the bandgap ↓ , the conductivity (σ) ↑ , and the activation energy (slope) ↓



- Improved metal oxides properties are necessary to enhance STH conversion efficiency in multijunction photoelectrodes (such as the UH Hybrid Photoelectrode)
- Tungsten trioxide is a good candidate but it only absorbs high energy photons. Its bandgap must be reduced.
- Doping via nitrogen or ammonia has proved to decrease the bandgap of reactively-sputtered tungsten trioxide films.
- Doping also enhances electronic properties of the film by increasing conductivity.
- This indicates that nitrogen doping (via NH_3 or N_2) increases the n-type characteristics of the film, introducing energy levels closer to the conduction band than to the valence band.



- The optical and electronic properties of the present hybrid-compatible metal oxides are key limiting factors to efficiency. This clearly defines the primary focus for continued research effort.
- By lowering the bandgap, doping of metal oxides might provide a pathway to enhance efficiency.
- Further study and film characterizations are necessary to understand the nature of the introduced states, and their effect on photoelectrochemical activity.

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