

A2: Material and Electrode Designs for Energy Storage and Conversion Battery Design and Application #<u>A2-0209</u>

Experimental diagnostic of Li-ion commercial cells, case studies : High power and High energy commercial GIC//LFP cells

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# **Objectives & Motivation**

From a previous study, Journal of Power Sources 194 (2009) 541–549 A peculiar behavior was observed while cycling a HE GIC//LFP cell:



We speculated it was the result of a combination of LLI and a "reservoir effect" arisen from the loading matching between the PE and the NE. As the discharge rate increases, a portion of the PE would not be fully utilized, freeing Li ions from the NE to compensate the LLI.

# **Objectives & Motivation**

Mechanistic degradation simulation agrees and suggests that will only occur for HE type cells, not HP.



Journal of Power Sources 219 (2012) 204-216

This study aims to provide experimental validation HE and HP cells cycled with specially designed protocol

Step protocol to access evolution of capacity loss for rates from C/25 to 2C  $\,$ 

Degradation mechanisms assessed from IC analysis and mechanistic emulation

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### Aging of the cells Capacity fading and rest cell voltages evolution



# Mechanistic emulation The 'alawa toolbox

### **Principles:**

OCV: 5 main degradation modes Loss of active materials On lithiated PE ( $LAM_{liPE}$ ) On delithiated PE ( $LAM_{dePE}$ ) On lithiated NE ( $LAM_{liNE}$ ) On delithiated NE ( $LAM_{deNE}$ ) Loss of lithium inventory (LLI)



Degradation modes affect thermodynamic properties differently:

$$\label{eq:lectrodes} Electrodes \ balance \qquad LR \ = \ LR_{ini} \bigg( \frac{100\% - \% LAM_{deNE} - \% LAM_{liNE}}{100\% - \% LAM_{dePE} - \% LAM_{liPE}} \bigg)$$

$$\label{eq:linear} \text{Lithium ions balance} \qquad \text{OFS} \,=\, \text{OFS}_{ini} + \text{LR} \times \text{\%LAM}_{liNE} - \frac{\text{LR}}{\text{LR}_{ini}} \times \text{\%LAM}_{dePE} + \text{\%LLI}$$

Unique approach Allows simulations of voltage response (IC,DV,RCVs), capacity and power fading

Details in Journal of Power Sources 219 (2012) 204-216

### Focus on the first 5% capacity loss for both cells HP cell: 350 cycles, HE cell: 150 cycles

Use of mechanistic emulation: the 'alawa toolbox Capacity loss (assuming linear degradation for each mode):



Simulations details in Journal of Power Sources 258 (2014) 408-419

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#### Look for other "in-situ sensors"

For HP cell, use resistance evolution

if 5% LAM<sub>liPE</sub>, resistance should increase (same current on smaller surface area) Resistance constant  $\rightarrow$  Capacity loss is due to loss of lithium inventory

#### For HE cell, use IC peak area evolution

SHAPE changed but peak area (i.e. the capacity of each reaction) is still relevant



### Focus on the first 5% capacity loss for both cells HE cell: 150 cycles – IC peak area evolution

### Combine modes to match capacity fade & area variations

LAM<sub>dePE</sub> and LAM<sub>liNE</sub> can occur without capacity loss BUT they'll alter voltage response If main LAM<sub>liNE</sub>: need to  $\uparrow ] \odot \& \uparrow ] \odot \odot$ No match If main LLI: need to  $\downarrow ] \odot \& \uparrow ] \odot \odot$ LAM<sub>deNE</sub> is matching





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# Focus on the HE cell – cycle 200 to 550 IC peak area evolution

### Same area analysis for the HE later cycles:

#### Additional LLI + trace of $LAM_{deNE}$



Mechanistic analysis proved that the capacity loss for both cells is solely induced by LLI and that it seems to be always accompanied by some  $LAM_{deNE}$ 

BUT it also raised more questions on the HE cell first 150 cycles:

Why at a 2.5x higher LLI pace? Why higher LAM<sub>deNE</sub> rate? Why did the kinetics improved (IC peak shape changed,  $Q_{C/2}$  and  $Q_{C/1}$  **7**)?

# Focus on the HE cell Origin of the higher LLI, LAM<sub>dene</sub> and kinetics

### Electrochemical milling of the PE?

#### Increases electrode surface area

Explain apparent resistance decrease



Explain increased capacity at high rates (because of high Peukert constant) Explain changes in peak shapes since PE kinetics improved

Grain cracking releases Fe ions that could have migrated to the NE Surface poisoning → Higher LLI and LAM

Electrochemical milling explains the additional features of the fist HE 150 cycles

### Experimental diagnostic of commercial HE & HP GIC//LFP cells Conclusions

Two commercial GIC//LFP cells were cycled and analyzed by electrochemical inference techniques and mechanistic emulations

The HP cell showed a linear loss of lithium inventory at about 0.016% per cycle associated with a 0.003% loss of active material on the NE.

The HE cell showed the same degradation but with some electrochemical milling in the first 150 cycles that induced better kinetics but additional 5% LLI and 5 % LAM<sub>deNE</sub>.

The use of mechanistic emulation and of the 'Alawa toolbox created unprecedented benefits to derive fading mechanism with quantitative comparisons for validation.

Full study : Journal of Power Sources 258 (2014) 408-419

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Mahalo for your attention! Questions?

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